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- (71) Applicant: AKZO NOBEL N.V. [NL/NL]; Velperweg 76,
NL-6824 BM Arnhem (NL).
- (72) Inventors: STAMIRE, Dennis; 6 Rockingham Drive,
Newport Beach, CA 92660 (US). JONES, William; 56
- (74) Agent: SCHALKWIJK, Peter, Cornelis; Akzo Nobel
N.V., Intellectual Property Department (Dept. AIP), P.O.
Box 9300, NL-6800 SB Arnhem (NL).
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(54) Title: IN SITU FORMED ANIONIC CLAY-CONTAINING BODIES

(57) Abstract: The present invention is directed to a process for the preparation of crystalline anionic clay-containing bodies from sources comprising a trivalent metal source and a divalent metal source comprising the steps of: a) preparing a precursor mixture containing a liquid, a divalent metal source and/or a trivalent metal source, at least one of them being insoluble in the liquid; b) shaping the precursor mixture to obtain shaped bodies; c) optionally thermally treating the shaped bodies; and d) aging the shaped bodies to obtain crystalline anionic clay-containing bodies; with the proviso that if no trivalent metal source is present in the precursor mixture of step a), such source is added to the shaped bodies after shaping step b) and before aging step d); and with the further proviso that the combined use of an aluminium source as the trivalent metal source and a magnesium source as the divalent metal source is excluded. The quintessence of the present invention is that the major part of the final amount of anionic clay is formed after shaping, i.e., in situ in the shaped body. This results in attrition resistant bodies, without the need to add a binder material.

IN SITU FORMED ANIONIC CLAY-CONTAINING BODIES

The present invention pertains to a process for the preparation of shaped crystalline anionic clay-containing bodies. Examples of crystalline anionic clays include hydrotalcite, meixnerite, sjögrenite, pyroaurite, stichtite, reevesite, eardleyite, manassite, and barbertonite. Crystalline anionic clays have several applications in the catalyst field and as absorbents. For most commercial applications crystalline anionic clays are formed into shaped bodies such as spheres. In all applications where shaped bodies are exposed to severe processing conditions and environments, such as oil refinery applications, separations, purifications, and absorption processes, it is of paramount importance that the integrity of the crystalline anionic clay-containing shaped bodies is kept intact and attrition is prevented.

In the prior art, crystalline anionic clays are usually incorporated into binder or matrix material in order to obtain attrition resistant shaped bodies. Commonly used binder or matrix materials are alumina, and silica. Frequently used alumina precursors are aluminium chlorohydrol, soluble aluminium salts, and acid dispersed pseudo-boehmite; regular silica precursors are silica sols, silicates, silica-alumina co-gels, and combinations thereof.

EP-0 278 535 describes FCC additives or catalyst particles which are prepared by embedding hydrotalcite and optionally zeolite in a silica, silica-alumina or alumina matrix. To this end, hydrotalcite is slurried in a matrix precursor dispersion containing the other catalyst components or precursors thereof and subsequently spray-dried.

However, when crystalline anionic clay is embedded in a matrix, the amount of active crystalline anionic clay contained within the resulting shaped bodies tends to be relatively small. There are applications in which performance

reasons it is desired that the shaped bodies consist or mostly consist of active crystalline anionic clay. Also, by the incorporation of crystalline anionic clay into matrix material, physical properties of the crystalline anionic clay such as specific surface area, pore size distribution, etcetera may be detrimentally affected. Further, the distribution of the crystalline anionic clay within the matrix is difficult to control. Another disadvantage of having to use a matrix to obtain attrition resistant bodies is the fact that most commonly used matrix/binder materials have some chemical activity, which in certain applications can cause undesirable side reactions. For instance, one of the most commonly used binder materials in FCC catalysts and additives is silica or silica based material. These types of binders are not suitable for use in sulphur oxides removal additives, because they detrimentally affect the sulphur removal.

The present invention provides crystalline anionic clay-containing bodies which are attrition resistant without high amounts of binder having to be present and/or added. In fact, crystalline anionic clay-containing bodies are provided which can be binder-free. The crystalline anionic clay distribution within the crystalline anionic clay-containing bodies of the invention can easily be controlled as will be explained further in the description. Within the context of the present specification the term "crystalline anionic clay" means a clay which has an X-ray diffraction pattern containing the specific X-ray diffraction pattern which characterises the specific type of anionic clay.

The present invention is directed to a process for the preparation of crystalline anionic clay-containing bodies from sources comprising a trivalent metal source and a divalent metal source comprising the steps of:

- a) preparing a precursor mixture containing a liquid, a divalent metal source and/or a trivalent metal source, at least one of them being insoluble in the liquid;

- b) shaping the precursor mixture to obtain shaped bodies;
 - c) optionally thermally treating the shaped bodies; and
 - d) aging the shaped bodies to obtain crystalline anionic clay-containing bodies;
- 5 with the proviso that if no divalent or trivalent metal source is present in the precursor mixture of step a), such source is added to the shaped bodies after shaping step b) and before aging step d);
- and with the further proviso that the combined use of an aluminium source as the trivalent metal source and a magnesium source as the divalent metal
- 10 source is excluded.

The quintessence of the present invention is that crystalline anionic clay is formed after shaping, i.e. in situ in the shaped body. This results in very attrition resistant bodies, without the need to add a binder material. In order to obtain a shaped body a solid precursor must be present in the precursor mixture and it is this context that at least one of the metal sources preferably is an oxide, a hydroxide, a carbonate, or a hydroxy carbonate.

- The shaped bodies can be prepared in various ways. In a preferred embodiment of the process a trivalent metal source and a divalent metal source are combined in a slurry to form a precursor mixture. Subsequently, said precursor mixture is shaped. The resulting shaped bodies are aged, optionally after thermal treatment, in a liquid to obtain crystalline anionic clay-containing bodies.
- 25 Optionally the precursor mixture is pre-aged prior to the shaping step. This may be advantageous, because during said pre-aging step nuclei may be formed which enhance the formation of the crystalline anionic clay during aging step d).

It is also possible to prepare a precursor mixture from only one source such as an oxide, hydroxide or carbonate of a trivalent metal source or a divalent metal source, shape it, and then add one or more additional other sources to the shaped bodies in any of the subsequent process steps. During the aging step, 5 the various sources react to give the crystalline anionic clay-containing bodies. It is also possible to use combinations of the two preparation routes described above, for instance: add the trivalent metal source and the divalent metal source to form the precursor mixture, shape to form bodies, and then age the shaped bodies in a liquid containing additional metal source to form anionic 10 clay-containing bodies with a higher content of said additional metal on the outside of the shaped body.

Suitable trivalent metals include aluminium, gallium, indium, iron, chromium, vanadium, cobalt, manganese, cerium, niobium, lanthanum. 15 Aluminium sources include aluminium alkoxide, aluminium oxides and hydroxides such as transition alumina, aluminium trihydrate (gibbsite, bayerite) and its thermally treated forms (including flash-calcined alumina), alumina sols, amorphous alumina, (pseudo)boehmite, aluminium-containing clays such as kaolin, sepiolite, and modified clays such as metakaolin, alumina salts such as aluminium nitrate, aluminium chloride, aluminium chlorohydrate, sodium aluminate, and aluminium sulphate. With the preparation method according to 20 the invention it is also possible to use coarser grades of aluminium trihydrate such as BOC (Bauxite Ore Concentrate) or bauxite. When clays are used as aluminium source it may be necessary to activate the 25 alumina in the clay by acid treatment - e.g. acid-treated bentonite -, base treatment, thermal treatment, hydrothermal treatment, or combinations thereof. Acid treatment comprises treatment with acids such as nitric acid, acetic acid, phosphoric acid, sulphuric acid, and hydrochloric acid. Thermal treatment is usually performed at temperatures ranging from 30°-1000°C, preferably 200°-

800°C, for a time ranging from several minutes to 24 hours, preferably 1-10 hours.

Suitable gallium, indium, iron, chromium, vanadium, cobalt, cerium, niobium, lanthanum and manganese sources are the respective oxides, hydroxides,

5 carbonates, nitrates, chlorides, chlorohydrates, and alkoxides.

Also mixtures of the above-mentioned trivalent metal sources can be used, or doped trivalent metal sources. Such doped metal sources are prepared by treatment of a trivalent metal source in the presence of an additive. An example of a doped trivalent metal source is doped boehmite.

10 If more than one trivalent metal source is used, these metal sources can be combined in the precursor mixture in any sequence.

It is also possible to add a trivalent metal source after the shaping step. In that case, the precursor mixture may or may not already contain a trivalent metal source. If a trivalent metal source is added after the shaping step, it preferably 15 is in liquid when contacted with the shaped bodies. This can be done by dispersing or dissolving the trivalent metal source and adding it to the shaped bodies.

Also other trivalent metal sources than clay, e.g. aluminium trihydrate, may be pre-treated prior to the addition to the precursor mixture or prior to contacting it 20 with the shaped bodies. Said pre-treatment may involve treatment with acid, treatment with base, thermal and/or hydrothermal treatment, all optionally in the presence of seeds.

It is not necessary to convert all of the trivalent metal source into crystalline anionic clay. For instance, any excess of aluminium source will be converted 25 into alumina (usually in the form of transition alumina such as γ -alumina or (crystalline) boehmite) during the aging step. These compounds improve the binding within the shaped bodies and may also impart additional desirable functionalities to the bodies. For instance, alumina provides acid sites for catalytic cracking and (crystalline) boehmite also improves the nickel 30 encapsulation capacity of the shaped bodies. The formation of (crystalline)

boehmite may be promoted by adding seeds, either in the precursor mixture, in the aluminium source or during aging.

Suitable divalent metal sources include magnesium, zinc, nickel, copper, iron,
5 cobalt, manganese, calcium, and barium.

Suitable magnesium sources are oxides or hydroxides such as MgO and
Mg(OH)₂, hydromagnesite, magnesium salts such as magnesium acetate,
magnesium formate, magnesium hydroxy acetate, magnesium carbonate,
magnesium hydroxy carbonate, magnesium bicarbonate, magnesium nitrate,
10 magnesium chloride, magnesium-containing clays such as dolomite, saponite,
and sepiolite. Suitable zinc, nickel, copper, iron, cobalt, manganese, calcium,
and barium sources are the respective oxides, hydroxides, carbonates, nitrates,
and chlorides.

Also mixtures of the above-mentioned divalent metal sources can be used, or
15 doped divalent metal sources. Such doped metal sources are prepared by
treatment of a divalent metal source metal source with a suitable dopant. An
example of a doped divalent metal source is doped brucite.

If more than one divalent metal source is used they can be combined in the
precursor mixture in any sequence and/or in any process step after the shaping
20 step. If a divalent metal source is added after the shaping step, it is preferably in
liquid when contacted with the shaped bodies. This can be done by dispersing
or dissolving the divalent metal source and adding it to the shaped bodies.

The divalent metal source may be pre-treated prior to the addition to the
precursor mixture and/or prior to the addition to the shaped bodies. Said pre-
25 treatment may comprise a thermal and/or a hydrothermal treatment, an acid
treatment, a base treatment, all optionally in the presence of a seed.

It is not necessary to convert all of the divalent metal source into crystalline
anionic clay. For instance, any excess of magnesium compounds will usually be
converted into brucite or magnesia. For the sake of clarity, this excess of
30 magnesium compounds in the shaped particle will be referred to in this

description as magnesia. The presence of magnesia in the shaped body may provide desirable functionalities to the shaped bodies such as for instance metal trap capacity. The presence of magnesia provides basic sites which render the shaped body suitable for treating strong acid streams of gases or liquids.
5 remove or neutralise undesirable acid components.

The shaped bodies containing anionic clay, anionic clay and magnesia or anionic clay and alumina may be used in processes involving purification and/or separation of organic compounds in hydrocarbon streams, e.g. the removal of
10 S-compounds and/or N-compounds in the gasoline and diesel fraction in FCC and hydroprocessing. Further, said shaped bodies may be used in water treatment for removing organic and inorganic compounds for the purpose of purifying, clarifying, and separating undesirable compounds from said water streams, including ion-exchange processes. Also, said shaped bodies may be
15 used in the treatment of gaseous streams in industrial processes to remove undesirable gaseous compounds such as chlorine, hydrochloric acid, sulphur compounds (e.g. SO_x), nitrogen compounds (e.g. NO_x, ammonia) and phosphorus compounds.

20 The various process steps will be described in more detail below.

Preparation of the precursor mixture

In this step a precursor mixture is prepared from a trivalent metal source and/or
25 a divalent metal source in a liquid. All liquids are suitable, as long as they do not detrimentally interfere with the various sources. Suitable liquids are water, ethanol, and propanol. The amount of liquid can be chosen such that a mixture with a milky substance is obtained, but also mixtures with a higher viscosity, for instance doughs, are suitable. If more than one source is used for the precursor
30 mixture, the sources can be added as solids, but they can also be added in

liquid, provided that the combination of an aluminium and a magnesium source is excluded. The various sources can be added in any sequence.

The preparation of the precursor mixture can be carried out with or without stirring, at room temperature or elevated temperature. Optionally, the precursor
5 mixture and/or the separate sources are homogenised by, for instance, milling, ultrasound treatment, or high shear mixing. Such treatments may also increase the reactivity and/or reduce the particle size of the metal source(s).

Some conversion to crystalline anionic clay may already take place upon combining the various sources. It is preferred that at least 5 wt% of the final
10 total amount of anionic clay is already formed, but for the present invention it is essential that conversion also takes place after the shaping step. Usually more than 25wt%, preferably more than 50 wt%, more preferably more than 75 wt% and most preferably between 80-95 wt% of the final amount of anionic clay in the shaped body is formed after the shaping step, because then shaped bodies
15 with the highest physical strength are obtained.

The divalent to trivalent metal molar ratio within the anionic clay may vary from 1 to 10, preferably 1 to 6, most preferably 2 to 4.

If desired, organic or inorganic acids and bases, for example for control of the pH, may be added to the precursor mixture or added to any one of the trivalent
20 metal source and/or divalent metal source before these are added to the precursor mixture. An example of a preferred modifier is an ammonium base, because upon drying no deleterious cations remain in the anionic clay.

As mentioned above, the precursor mixture may be pre-aged prior to the shaping step. The pre-aging temperature may range from 30° to 500°C and it
25 may be conducted under atmospheric or increased pressure such as autogeneous pressure at temperatures above 100°C. The aging time can vary from 1 minute to several days, for instance 7 days.

By adding specific anions to the precursor mixture and/or any of the trivalent metal and/or divalent metal sources the interlayer-charge balancing anions
30 present may be controlled. Usually, the pH must be controlled to introduce the

desired form of the interlayer charge balancing anion, many charge balancing anions being pH-dependent. Examples of suitable anions are carbonate, bicarbonate, nitrate, chloride, sulphate, bisulphate, vanadates, tungstates, borates, phosphates, pillararing anions such as HVO_4^{4-} , $\text{V}_2\text{O}_7^{4-}$, $\text{HV}_2\text{O}_{12}^{4-}$, $\text{V}_3\text{O}_9^{3-}$,
5 $\text{V}_{10}\text{O}_{28}^{-6}$, $\text{Mo}_7\text{O}_{24}^{6-}$, $\text{PW}_{12}\text{O}_{40}^{3-}$, $\text{B}(\text{OH})_4^-$, $[\text{B}_3\text{O}_3(\text{OH})_4]^-$, $[\text{B}_3\text{O}_3(\text{OH})_5]^{2-}$,
 $\text{B}_4\text{O}_5(\text{OH})_4^{2-}$, HBO_4^{2-} , HGaO_3^{2-} , CrO_4^{2-} , and Keggin-ions, formates, acetate, and mixtures thereof. It is believed that the presence of some of anions such as carbonate, bicarbonate, sulphate and or nitrate influences the formation of side products such as brucite. Further, the addition of ammonium hydroxide
10 promotes meixnerite formation, whereas the addition of ammonium carbonate promotes hydrotalcite formation. It goes without saying that in those cases wherein a specific anion is preferred in the anionic clay the reaction conditions in the further preparation steps must be adapted so as to avoid exchange of the anion by other less preferred anions.

15

Shaping

Suitable shaping methods include spray-drying, pelletising, granulation, extrusion (optionally combined with kneading), beading, or any other
20 conventional shaping method used in the catalyst and absorbent fields or combinations thereof. The amount of liquid present in the precursor mixture should be adapted to the specific shaping step to be conducted. To this end one may partially remove the liquid used in the precursor mixture and/or add additional or other liquid, and/or change the pH of the precursor mixture to
25 make the precursor mixture gellable and thus suitable for shaping. Various additives commonly used in the various shaping methods such as extrusion additives may be added to the precursor mixture used for shaping.

Thermal treatment

After shaping the shaped bodies may optionally be submitted to a thermal treatment. Such a treatment increases the physical strength of the particles.

- 5 The thermal treatment can be conducted in an oxygen-containing atmosphere, a hydrogen-containing atmosphere, in an inert atmosphere or in steam at temperatures varying from 30° to 900°C for a time ranging from a few minutes to 24 hours. As in, for instance, spray-drying a thermal treatment is inherently involved, a further thermal treatment may not be necessary.

10

Aging

- In this step, the shaped bodies are immersed in a protic liquid or protic gaseous medium. During the aging step crystallisation to crystalline anionic clay takes place. Suitable protic liquids or gaseous media are those in which the shaped bodies do not dissolve, such as water, ethanol, methanol, propanol, steam, gaseous water, and gaseous ethanol. Increasing the temperature and/or the pressure can reduce the aging time. Aging can be conducted under autogeneous conditions. The aging temperature may range from 30° to 500°C.

- 15 20 The aging time can vary from 1 minute to several days, for instance 7 days. For some purposes it is advantageous to conduct several aging steps, optionally with intermediate drying steps, optionally followed by calcination steps. For instance, an aging step at a temperature below 100°C may be followed by a hydrothermal aging step at a temperature above 100°C and autogeneous pressure, or vice versa.

- 25 As will be described below in further detail, additives can be added before, after or during any aging step. By adding specific anions to the aging medium at controlled pH, the interlayer-charge balancing anions present can be controlled. Examples of suitable anions are carbonate, bicarbonate, nitrate, chloride, sulphate, bisulfate, vanadates, tungstates, borates, phosphates, pillarating anions

- such as HVO_4^- , $\text{V}_2\text{O}_7^{4-}$, $\text{HV}_2\text{O}_{12}^{4-}$, $\text{V}_3\text{O}_9^{3-}$, $\text{V}_{10}\text{O}_{28}^{-6}$, $\text{Mo}_7\text{O}_{24}^{6-}$, $\text{PW}_{12}\text{O}_{40}^{3-}$, B(OH)_4^- , $[\text{B}_3\text{O}_3(\text{OH})_4]^-$, $[\text{B}_3\text{O}_3(\text{OH})_5]^{2-}$, $\text{B}_4\text{O}_5(\text{OH})_4^{2-}$, HBO_4^{2-} , HGaO_3^{2-} , CrO_4^{2-} , and Keggin-ions, formate, acetate, and mixtures thereof. It is also believed that the presence of some of these anions such as carbonate, bicarbonate, sulphate, 5 and/or nitrate influence the forming of side products such as brucite. Further, the addition of ammonium hydroxide promotes meixnerite-like clay formation, whereas the addition of ammonium carbonate promotes hydrotalcite-like clay formation.
- 10 For some applications it is desirable to have additives present in and/or on the shaped bodies according to the invention. Suitable additives include compounds of rare earth metals (especially Ce and La), Si, P, B, Bi, Group VI metals, Group VIII metals, noble metals such as Pt and Pd, alkaline earth metals (for instance Ca and Ba) and/or transition metals (for example Mn, Fe, 15 Ti, V, W, Zr, Cu, Ni, Zn, Mo, Sn). The additives or their precursors can be added separately or in mixtures in any of the preparation steps of the invention. For instance, they can be deposited on the shaped bodies before, during, or after aging, or else they can be added to the precursor mixture and/or any of the trivalent metal or divalent metal sources. Suitable sources of metal compounds 20 and non-metal compounds are oxides, halides such as chlorides, sulfates, nitrates, and phosphates. As mentioned above, the additives may be added in any of the preparation steps. This can be especially advantageous for controlling the distribution of the additives in the shaped bodies. It is even possible to calcine the shaped bodies, rehydrate them in the presence of 25 anions, such HVO_4^- , $\text{V}_2\text{O}_7^{4-}$, $\text{HV}_2\text{O}_{12}^{4-}$, $\text{V}_3\text{O}_9^{3-}$, $\text{V}_{10}\text{O}_{28}^{-6}$, $\text{Mo}_7\text{O}_{24}^{6-}$, $\text{PW}_{12}\text{O}_{40}^{3-}$, B(OH)_4^- , $[\text{B}_3\text{O}_3(\text{OH})_4]^-$, $[\text{B}_3\text{O}_3(\text{OH})_5]^{2-}$, $\text{B}_4\text{O}_5(\text{OH})_4^{2-}$, HBO_4^{2-} , HGaO_3^{2-} , CrO_4^{2-} , Keggin-ions, formate, acetate, and mixtures thereof. It is further possible to reduce, hydrogenate or sulphurize the metals after and/or during introduction. With the help of additives the shaped bodies may be provided with desired 30 functionalities, or the desired functionality may be increased by the addition of

- additives. The suitability of anionic clay-containing shaped bodies for the removal of SO_x and/or NO_x compounds in FCC may be improved by the addition of Ce and/or V. The presence of V, W, Mo and/or Zn improves the suitability for removal of S-compounds in the gasoline and diesel fraction of
- 5 FCC. The presence of Zn and/or Mn improves metal trapping. As described above, these functionalities may also be built in by using an excess of trivalent metal source and/or divalent metal source. A combination of these measures increases the effect.
- 10 The crystalline anionic clay-containing bodies may also be prepared to contain conventional catalyst components such as matrix or filler materials (e.g. kaolin clay, phosphated kaolin, titanium oxide, zirconia, alumina, silica, silica-alumina, and bentonite), and molecular sieves (e.g. zeolite Y, USY zeolite, ion-exchanged zeolite, ZSM-5, beta-zeolite, and ST-5). These conventional catalyst
- 15 components may be added prior to the shaping step. Because the anionic clay is formed in situ, the resulting body will have a homogeneous dispersion of anionic clay and catalyst components. With the method according to the invention, multiple-functional bodies can be prepared which can be used as a catalyst or as a catalyst additive.
- 20 The process according to the invention may be conducted batch-wise or in a continuous mode, optionally in a continuous multi-step operation. The process may also be conducted partly batch-wise and partly continuous.
- 25 If desired, the crystalline anionic clay-containing shaped bodies prepared by the process according to the invention may be subjected to ion-exchange, in which the interlayer charge-balancing anions of the clay are replaced with other anions. Said other anions are the ones commonly present in anionic clays and include pillaring anions such as HVO₄⁴⁻, V₂O₇⁴⁻, HV₂O₁₂⁴⁻, V₃O₉³⁻, V₁₀O₂₈⁶⁻,
- 30 Mo₇O₂₄⁶⁻, PW₁₂O₄₀³⁻, B(OH)₄⁻, [B₃O₃(OH)₄]⁻, [B₃O₃(OH)₅]²⁻, B₄O₅(OH)₄²⁻, HBO₄²⁻

HGaO₃²⁻, CrO₄²⁻, and Keggin-ions. Examples of suitable pillaring anions are given in US 4,774,212, which is included by reference for this purpose. Said ion-exchange can be conducted as soon as the crystalline anionic clay has been formed.

5

The present invention is further directed to crystalline anionic clay-containing shaped bodies obtainable by the process according to the invention. As mentioned above, the shaped bodies appear to have high mechanical strength and attrition resistance, comparable with that of clay-containing bodies prepared 10 by dispersing clay into a matrix or binder material and then shaping the clay-containing composition, yet without matrix or binder material having to be added to the crystalline anionic clay. This means that with the process according to the invention, crystalline anionic clay-containing shaped bodies can be prepared containing more than 25 wt%, preferably more than 50 wt%, more preferably 15 more than 70 wt% or even more preferably more than 90 wt% crystalline anionic clay. Although binder material may be present in the shaped crystalline anionic clay containing bodies according to the invention, for instance as a result of an excess aluminium source present in the precursor mixture, any binder that is present in the shaped bodies according to the invention will be 20 present as a discontinuous phase, as is depicted schematically in Figure 1. This is in contrast to clay-containing bodies prepared in the conventional way, i.e. by embedding clay into a matrix or binder material, where the binder material in the bodies is present in a continuous phase, as is schematically depicted in Figure 2. It is of course also possible to incorporate the crystalline anionic clay-25 containing shaped bodies into a matrix. In that case, composite particles are obtained which comprise crystalline anionic clay-containing shaped bodies with, optionally, binder material in a discontinuous phase, embedded in a binder material, as is schematically depicted in figure 3.

Figure 1: Schematic view of a shaped body according to the invention

Figure 2: Schematic view of a shaped body according to the prior art

Figure 3: Schematic view of a composite particle comprising shaped bodies according to the invention

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In Figure 1 a schematic view is given of a shaped crystalline anionic clay-containing body (1) according to the present invention which comprises crystalline anionic clay (2) and binder material (3) in a discontinuous phase.

10 In Figure 2 a schematic view is given of a shaped crystalline anionic clay-containing body (1) according to the prior art which comprises crystalline anionic clay (2) and binder material (3) in a continuous phase.

15 In Figure 3 a schematic view is given of a composite particle comprising crystalline anionic clay-containing shaped bodies (1) which comprise crystalline anionic clay (2) and binder material (3) in a discontinuous phase which are embedded in binder material (3') which is in a continuous phase.

During or prior to use in catalytic applications, anionic clays are often thermally treated to obtain so-called solid solutions or spinels. The present invention is also directed to shaped bodies and composite particles containing thermally 20 treated anionic clay.

The invention is illustrated by the following Examples.

EXAMPLES

25

Example 1

Flash-calcined gibbsite, Cp grade, was slurried in water that contained zinc carbonate. The Zn:Al atom ratio was 2. The slurry was homogenized with shear 30 mixing. The slurry was filtered and the filter cake was granulated to form shaped

bodies. The shaped bodies were calcined at 250°C for 4 hours. The calcined shaped bodies were slurried in water and aged at 65°C for 6 hours. The pH of the slurry was adjusted to 6.5 with nitric acid: XRD analysis indicated the presence of Zn-Al hydrotalcite and some ZnO in the shaped bodies.

5

Example 2

Flash-calcined gibbsite, Cp grade, was slurried in water that contained iron (II) nitrate. The slurry was homogenized with shear mixing. The slurry was filtered 10 and the filter cake was granulated to form shaped bodies. The shaped bodies were calcined at 250°C for 4 hours. The calcined shaped bodies were slurried in water and aged at 65°C for 18 hours. The pH of the slurry was adjusted to 9.5 with ammonium hydroxide. XRD analysis indicated the presence of Fe-Al hydrotalcite in the shaped bodies.

15

Example 3

Gallium nitrate was added to an aqueous slurry containing magnesium oxide. The slurry was homogenized with shear mixing and formed to shaped bodies by 20 spray-drying. The shaped bodies were calcined at 250°C for 4 hours. The calcined shaped bodies were slurried in water and aged at 65°C for 18 hours. The pH of the slurry was adjusted to 9.5 with ammonium hydroxide. XRD analysis indicated the presence of Mg-Ga hydrotalcite in the shaped bodies.

25

Example 4

Aluminium trihydrate (46.5) was slurried in 466 g de-ionised water containing 389.6 g $\text{Fe}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$. The total amount of slurry was 976 g and had a solids content of 13 wt%. The resulting slurry was aged and the resulting product was 30 granulated. The granules were aged hydrothermally at 175°C for 2 hours. The

product was dried overnight at 110°C. XRD showed the formation of Fe-Al anionic clay.

Example 5

5

Example 1 was repeated, except that cobalt nitrate being used instead of zinc carbonate. The process conditions were the same. The PXRD pattern of the final product indicated the formation of a Co-Al anionic clay.

10 Example 6

Ferric hydroxide was prepared by precipitation from a ferric nitrate solution. Ferrous hydroxide was precipitated from a ferrous nitrate solution by addition of ammonium hydroxide in an inert atmosphere of nitrogen. The two precipitates 15 were combined and shear mixed. One half of the mixture was aged at 85°C for 8 hours in a closed container. The other half was aged at 150°C for 30 minutes. Both mixtures were filtered and the filter cake was granulated into shaped bodies. The shaped bodies were calcined at 200°C for 4 hours and then rehydrated at 65°C in water for 6 hours. The products were dried at 110°C. 20 PXRD showed the formation of a $\text{Fe}^{3+}\text{Fe}^{2+}$ -anionic clay in both products.

CLAIMS

1. Process for the preparation of crystalline anionic clay-containing bodies from sources comprising a trivalent metal source and a divalent metal source comprising the steps of:
 - 5 a) preparing a precursor mixture containing a liquid, a divalent metal source and/or a trivalent metal source, at least one of them being insoluble in the liquid;
 - b) shaping the precursor mixture to obtain shaped bodies;
 - 10 c) optionally thermally treating the shaped bodies; and
 - d) aging the shaped bodies to obtain crystalline anionic clay-containing bodies;

with the proviso that if no divalent or trivalent metal source is present in the precursor mixture of step a), such source is added to the shaped bodies after shaping step b) and before aging step d);

15 and with the further proviso that the combined use of an aluminium source as the trivalent metal source and a magnesium source as the divalent metal source is excluded.
- 20 2. Process according to claim 1, wherein the precursor mixture contains a divalent metal source and a trivalent metal source.
3. Process according to claim 1 or 2 wherein the precursor mixture is pre-aged prior to shaping step b).
- 25 4. Process according to claim 2 or 3 wherein in step a) a trivalent metal source and a magnesium source are combined to obtain a precursor mixture.

5. Process according to any one of claims 1-4 wherein the trivalent metal source is present in the precursor mixture and a divalent metal source is added after the shaping step b) and wherein the trivalent metal source is
5 selected from the group consisting of oxides, hydroxides, carbonates, hydroxy carbonates and combinations thereof.
10. Process according to any one of claims 1-5 wherein a divalent metal source is present in the precursor mixture and a trivalent metal source is added after the shaping step b) and wherein the divalent metal source is selected from the group consisting of oxides, hydroxides, carbonates, hydroxy carbonates and combinations thereof.
15. Process according to any one of claims 1-6 wherein the trivalent metal source is selected from aluminium trihydrate, its thermally treated form or boehmite.
20. Process according to any one of the claims 1-7 wherein the trivalent metal source comprises kaolin, phosphated kaolin, bentonite, metakaolin and/or bauxite.
25. Process according to any one of claims 1-6 wherein the divalent metal source comprises magnesium oxide.
30. 10. Process according to any one of the preceding claims wherein more than one aging step is conducted, optionally with intermediate drying steps, optionally followed by calcining.
11. Process according to any one of the preceding claims wherein additives are added in step a).

12. Process according to any one of the preceding claims wherein additives are added after the shaping step b).
13. Process according to any one of the preceding claims wherein additives are
5 added in any one of the aging steps.
 ii
14. Crystalline anionic clay-containing shaped body obtainable by any one of the processes of claims 1-13.
- 10 15. Crystalline anionic clay-containing body wherein any binding material present in the body is present in a discontinuous phase.
 ii
16. Crystalline anionic clay-containing body according to claim 14 or 15 wherein alumina is present.
 ii
- 15 17. Crystalline anionic clay-containing body according to any one of claims 14-16 wherein magnesia is present.
18. Shaped body containing thermally treated anionic clay obtained by thermally
20 treating a crystalline anionic clay-containing body according to any one of claims 14-17.
19. Composite particle comprising a crystalline anionic clay-containing body according to any one of claims 14-17 which is embedded in binder material.
- 25 20. Composite particle comprising a shaped body containing thermally treated anionic clay according to claim 18.

21. Process for the purification and/or separation of organic compounds in a hydrocarbon stream wherein a crystalline anionic clay-containing body according to any one claims 14-17 is used.
- 5 22. Process for the removal of organic and inorganic compounds for the purpose of purifying, clarifying or separating undesirable compounds from water wherein a crystalline anionic clay-containing body according to any one of claims 14-17 is used.
- 10 23. Process for the removal of undesirable gaseous components from a gaseous stream wherein a crystalline anionic clay-containing body according to any one of claims 14-17 is used. {1}

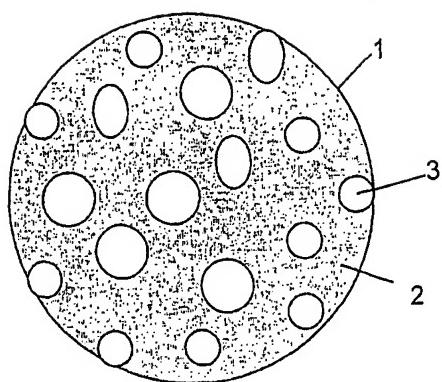


Figure 1. Shaped body according to the invention

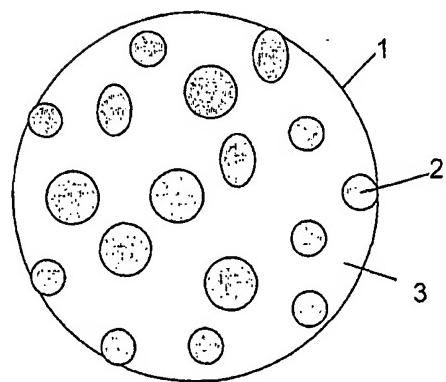


Figure 2. Shaped body according to the prior art

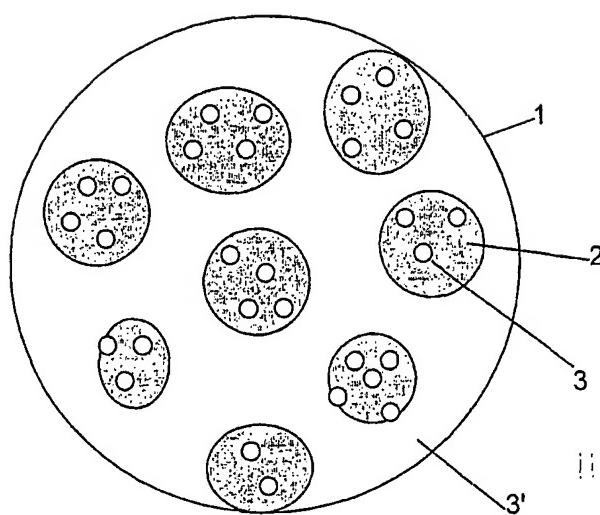


Figure 3. Composite particle comprising shaped bodies according to the invention

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 02/01110

A. CLASSIFICATION OF SUBJECT MATTER			
IPC 7	C01F7/00	C01G15/00	B01J21/16
			C01B13/32

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C01F C01G B01J C01B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the International search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ, COMPENDEX, INSPEC, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 278 535 A (AKZO NV) 17 August 1988 (1988-08-17) cited in the application page 7, line 40 -page 8, line 50 ---	14-16, 18-21,23
A	US 4 454 244 A (WOLTERMANN GERALD M) 12 June 1984 (1984-06-12) examples 1,2 ---	1-23
P, X	WO 01 12319 A (AKZO NOBEL NV) 22 February 2001 (2001-02-22) page 2, line 20 -page 3, line 5; claims 1,22 page 9, line 14-19 ---	14-16, 18-21,23 -/-

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

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- *&* document member of the same patent family

Date of the actual completion of the international search

17 July 2002

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Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL-2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Raming, T

INTERNATIONAL SEARCH REPORT

International Application No
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C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	LOPEZ-SALINAS ESTEBAN ET AL: "New gallium-substituted hydrotalcites: 'Mg1-xGax(OH)2! (CO3)x/2 DOT mH2O'" J POROUS MATER; JOURNAL OF POROUS MATERIALS 1996 KLUWER ACADEMIC PUBLISHERS, DORDRECHT, NETHERLANDS, vol. 3, no. 3, 1996, pages 169-174, XP008006008 page 170, left-hand column, paragraph 2 -right-hand column, paragraph 1 -----	1-23

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 02/01110

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
EP 0278535	A	17-08-1988	AT AU AU BR CA DE EP GR JP JP US US US	60874 T 609763 B2 1021288 A 8707099 A 1314862 A1 3861771 D1 0278535 A1 3001934 T3 2678276 B2 63182031 A 4866019 A 4946581 A 4952382 A	15-02-1991 09-05-1991 14-07-1988 02-08-1988 23-03-1993 21-03-1991 17-08-1988 23-11-1992 17-11-1997 27-07-1988 12-09-1989 07-08-1990 28-08-1990
US 4454244	A	12-06-1984	AT CA DE DE EP JP JP JP	79837 T 1209785 A1 3485888 D1 3485888 T2 0131685 A2 1628367 C 2049243 B 59207806 A	15-09-1992 19-08-1986 01-10-1992 04-03-1993 23-01-1985 20-12-1991 29-10-1990 26-11-1984
WO 0112319	A	22-02-2001	BR WO WO EP	0013132 A 0112545 A1 0112319 A1 1204474 A1	30-04-2002 22-02-2001 22-02-2001 15-05-2002